

Tanta University **Faculty of Science Department of Chemistry**

Final Examination for Second and Third Level Students of Chemistry/ Geology, Zoology, Microbiology, Botany, Entomology. Biochemistry)

Chemical Kinetics Chemistry

Course code: CH 2240

June 13th, 2015

Term: Second Total Assessment Marks: 100 Time Allowed: 2h

Answer all the following questions:

(20 marks for each)

- 1- a) Discuss the factors affecting the reaction rate?
 - b) An elementary reaction $2A + C \rightarrow D$, is second order in A and first order in C and the rate of this reaction is 2.5×10^{-1} M/S .when the concentration of A. C and D are all 1.0 M. What is the rate constant of this reaction?
- 2- a) Define the rate equation of chemical reaction and discuss how can you determine it?
 - b) The following data were obtained in the decomposition of $\rm\,N_2O_5$ in CCL4 at 40 CO

t (sec)	600	1200	1800	2400	3000	ω	
$O_2(ml)$	6.30	11.40	15.53	18.90	21.70	34.75	

Find out the order of this reaction and its half life time?

- 3- a) Enumerate the methods for determination the order of chemical reaction and discuss the differential method?
 - b) The half-life for radioactive disintegration of radium is 1590 Yr. calculate the decay constant .In how many years will three-quarter of the radium have undergone decay? (The decay is first order)

(انظر خلفه)

4- a) Deduce the integrated rate equation of the opposing first-order reaction

$$A \stackrel{\mathsf{K_I}}{\Longleftrightarrow} B$$

- b) The half-life of thermal denaturation of hemoglobin first order process has been to be 3460 Sec at 60 c° and at 65 c° . Calculate the activation energy (ΔE).
- 5- a) Write short notes about characteristics and classifications of catalysts and discuss the mechanism of chemical catalysis according to Arrhenius concept (Equilibrium treatment).
 - b) The reaction mechanism, $A + B \xrightarrow{k_1} C \xrightarrow{k_2} P$ if k >> k. Find out the rate law and the order of this reaction.

GOOD LUCK

Prof. Dr M. Y. EL SHEKH

Prof. Dr Hosny EL-Daly



			TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF CHEMISTRY		
		FINAL E	EXAM FOR LEVEL2 STUDENTS (DOUB)	LE MAJOR)	
1968	COURSE TITLE:	KINETIC THEORY OF GASES COURSE		COURSE CODE: CH2242	
^ DATE	JUNE10, 2015	TERM: SECOND TOTAL ASSESSMENT MARKS: 50 TIME ALLOWED		TIME ALLOWED: 2H	

Answer All Questions (50 marks)

1- Put true or false sign and correct the false answer? (10 marks)

- 1. Heavier gas molecules move faster than lighter molecules.
- 2. The gas viscosity increases as the pressure increases.
- 3. Collision cross section is the average distance a molecule travels between collisions.
- 4. A diatomic molecule has 5 degrees of freedom.
- 5. The mean free path decreases as the gas density increases.

2- Complete the following sentence: (20 marks)

- 1. Effusion of gases is defined as
- 2. All gases at a given temperature have the same
- 3. According to the equipartition of energy principle the average energy for each vibrational degree of freedom is......
- 4. Increasing the number of gas molecules will the mean free path.
- 5. For triatomic molecule the average total energy per mole is.......
- 6. is the number of variables required to describe the motion of a particle completely.
- 7. When the two gas molecules are headed in the same direction the consequent collision is called.....

Turn over the paper

- 8. The viscosity of gases increases as temperature increases. This is because
- 10. The molecular collisions are assumed to be elastic this means......

(4)- Write the mathematical equations (define its parameters) describes the following: (14 marks)

- 1. Mean free path
- 2. The collision frequency if each sphere is collides with the other.
- 3. Heat capacity at constant pressure for a gas.
- 4. Poiseuille's equation
- 5. The dependency of the barometric pressure on the altitude.
- 6. The volume excluded per mole of gas.
- 7. The relation between the thermal conductivity coefficient of a gas and the viscosity of a gas.

(5)- Answer the following? (6 marks)

- 1. If equal amounts of helium and hydrogen are placed in a porous container and allowed to escape, which gas will escape faster and how much faster?
- 2. Use the van der Waals equation to calculate the pressure of a sample of 5 moles of oxygen gas in a 10 L vessel at 10° C wherea $(O_2) = 1.36 \text{ L}^2 \text{ atm/mol}^2 \text{ and b } (O_2) = 0.032 \text{ L/mol}.$

Best Wishes ...

Dr. Eman Fahmy Aboelfetoh

TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF ZOOLOGY

EXAMINATION FOR SOPHOMOR (SECOND YEAR) STUDENTS OF ENTOMOLOGY AND CHEMISTRY

COURSE TITLE

INSECT INTERRELATIONSHIPS WITH OTHER LIVING **ORGANISMS**

COURSE CODE:EN 2242

on....

DATE: JUNE, 2015 TERM: SECOND TOTAL ASSESSMENT MARKS:150 TIME ALLOWED: 2 HOURS
PLEASE NOTE THE EXAM IS IN TWO (2) PAGES
Answer the following questions:
The First Question (Total 37½ Marks)
A. "Human exposure to sensitizing insect antigens causes immunological reactions ". Explain, giving examples whenever possible
B. Complete the following sentences
 C. Indicate whether the following statements are <u>True</u> or <u>False</u>
The Second Question (Total 37½ Marks)
 A. Choose from the brackets the correct answer and rewrite it in your paper (10 Marks, 2 Marks each) The colony of honey hees contains (Two queensOnly one queen Several queens). The relationship between the ants and aphids is defined as (cannibalismsymbiosisparasitism). Female (Ear wig Stink bugAcromis sparsa) stays with her offspring until they reach adulthood. The female of (PentatomidaeBelostomatidaeCarabidae) lays her eggs on the back of the male after mating.
5) (AphidsWaspsLocusts)live in social life B. Fill in the blanks with the appropriate words
 C. Correct the following statements and rewrite in your paper
4) The female chrysomelid beetle, Acromis sparsa, just lays eggs and does not care. 5) Predators live in or on the bodies of their insect host D. Write short notes on: 1) Two examples of insect predators
The Third Question (Total 37½ Marks)
A. Complete the following sentences

	4)	is an example of bacteria that cause diseases toinsects.
	5)	is an example of fungi that cause disease to house fly in which the dead flies are
D	Ch	surrounded by
D.	41	cose from brackets the correct answer and rewrite it in your paper
	2)	The muscardine disease of insects is caused by
	3)	The wilt disease of silk worm is caused by
	4)	Potato tip burn caused by the feeding of (Leafhoppers – <i>Gryllotalpa</i> – <i>Ceratitis capitata</i>) is a plant disease transmitted by insects and caused by the fungus <i>Ovulinia azalea</i> .
	4)	(Azalea flower spot – Sooty mold – Potato black leg)
	5)	Brown rot of apple is a plant disease transmitted by insects and caused by
	,	(Sclerotinia fructigena – Claviceps purpurea – Ovulinia azalea)
	6)	Ergot is aplant disease transmitted by insects. (Fugal – Bacterial – Viral)
	7)	is a plant disease transmitted by insects through oviposition wounds
	8)	(Tree cricket cancer of apples – Brown rot of apples)is an example of plant bacterial disease transmitted by insects.
	-	(Potato blackleg – Muscardine – Wilts disease)
	9)	is a plant viral disease transmitted by insects (Potato leaf roll – Banana bunchy top – Both)
0	Die) Aphids transmit virus from infected tobacco to healthy tobacco (Mosaic – Leaf roll – Bunchy top)
C.	DIS	cuss the following items
		The main differences between parasitism and predation.
		Mechanism of feeding of Entomophagous plants.
	3)	Symbiosis between acacia plants and ants.
		The Fourth Question (Total 37½ Marks)
A.	Ind	
Α.	Ind 1)	icate whether the following statements are <u>True</u> or <u>False</u>
Α.	1)	icate whether the following statements are <u>True</u> or <u>False</u>
Α.	1)	icate whether the following statements are <u>True</u> or <u>False</u>
A.	1)	icate whether the following statements are <u>True</u> or <u>False</u>
	2)	Symbiotic bacteria residing in the hindgut chambers of scarab beetle larvae may be useful in paratransgenic approaches to reduce larval root-feeding activities on agricultural crops. Nuclear polyhedral viruses affect about 200 species of Lepidoptera and the infected insect usually die. The ciliate <i>Tetrahymena dimorpha</i> occurs in the gut epithelium of larvae, pupae and adults.
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GOOD LUCK

EXAMINERS	PROF. IBRAHIM BAKR HELAL	PROF. SAID NOR EL-DEEN
	DR. SAMAR EZZAT ELKHOLY	DR. IMAN M. ELHUSSEINY

DEPARTMENT OF ZOOLOGY	
EXAMINATION PAPER FOR FRESHMEN (2ND YEAR) ST	The state of the s
COURSE TITLE: INSECT ECOLOGY	COURSE CODE: EN 2244
TE: 6/6/2015 JUNE, 2015 TERM:SECOND TOTAL MARKS: 150	TIME ALLOWED: 2 HOURS
rt 1 l in the blanks with the appropriate words (10 marks): Biodiversity is within a given ecosystem, bio The future depends on ecological understanding and our abilit	ome, or an entire planet. Ty to under differe
scenarios	
The term Ecosystem emphasizes that	• • • • • • • • • • • • • • • • • • • •
Much of the deviation among populations isolated on islands ap	ppears to be due to
The fittest individuals tend to and reproduce better.	
Indicate whether the following statements true or false and corre Insets, as poikilothermic animals, maintain a constant body tem temperature of the surroundings. Deserts are the most productive of the Earth's biomes. The rain probably exerts its influence on most insect population and quality of food or the incidence of disease. Mostly, circadian rhythms are endogenous, but are regulat factors. The fittest individuals tend to survive and reproduce better.	nperature irrespective to the
Choose the correct answer (10 marks): The organisms that breed together in nature to produce fertile (a) similar. b) different. c) isolated. d) all above mentions Synchronized eclosion increases the chances to	oned.
a) find a mate and food. b) escape potential predat	
c) find oviposition sites. d) all above mentioned.	
Cold hardiness refers to an insect's ability to adapt to and survi	
a) low temperature. b) high temperature. c) moderate tempe	erature.
The study of the interrelationships of living organisms and their	
a) Ecology b) habit c) biotic	
The number of individuals of a species of organism in a given an	rea at a particular time is th
a) community b) population c) habitat	

Part 2

1.

5.	Complete	the	following	sentences	(24 marks)
00	Complete	tille	Inmoning	Sentences	(24 marks)

match to their environments.

1-The ability to extract nutrients from the various textures is primarily due to.....

3. Define Ecotype giving an example of the recent publications that showed how ecotypes

2- In endo-parasitoids, the ovipositing females are able to distinguish between.....

3-In most communities; insects occupy an intermediate positions along food chain acting as 1, 2, 3, 4
 When new species are formed in isolated peripheral populations; which are isolated and prevented from exchanging genes, speciation is a) peripatric. b) allopatric. c) sympatric. Polymorphism is the occurrence together in the same habitat of two or more forms of species.
a) discontinuous. b) continuous. c) both. 3- Insects generate a great deal of complexity in food web atlevel. a) carnivore b) herbivore c) decomposer 4. Insects actually constitute thelevel after microorganisms except termites a) third b) second c) first 5-In Paedogenesis the immature stages produce a progeny. Without a) feeding b) suitable conditions c) fertilization 6- In the nitrogen cycle, the transformation of gaseous nitrogen into nitrogen- containing compounds is performed primarily by a) Fugi b) bacteria c) green plants d) herbivores
Write short note on: (16 marks) 1-Conservation of water by insects. (4 marks) 2-The main function of the filter- chamber in Homoptera. (6 marks) 3. Nitrification – ammonification. (6 marks)
Part 3 I) Define the following (6 marks) Community – Disturbance – Species richness.
II) What are the basic patterns of dispersion? If 5 quadrates had 10, 14, 9, 10 and 12 individuals, what type of dispersion would this be calculated? (6 marks)
III) Compare between exponential and Geometric population growth? (10 marks)
 IV) Choose the correct answer (28 marks) Which of the following could be included in a grasshopper niche, but not a description of its habitat?
a) continent where it lives. b) plant species it eats. c) location where it shelters. a) Intraspecific predation b) interspecific predation c) competition

AT.	TANTA UNIVERSITY FACULTY OF SCIENCE				
		DEPARTMENT OF ZOOLOGY EXAMINATION PAPER FOR FRESHMEN (2 ND YEAR) STUDENTS OF CHEM/ENT			
		AMINATION PAPER		YEAR) STUDENTS	
DATE MANAGEMENT	COURSE TITLE:		INSECT ECOLOGY		COURSE CODE: EN 2244
DATE: 6/6/2015	JUNE, 2015	TERM:SECOND	TOTAL MARKS: 1	50	TIME ALLOWED: 2 HOURS
 Biodivers The futur scenarios The term 	Ecosystem emp	withicological under	n a given ecosys	ur ability to	
5. The fittes	t individuals ter	nd to	. and reproduce	e better.	to be due to
 Insets, as temperate Deserts and The rain pand quality Mostly, of factors. 	poikilothermic are of the surro re the most prod probably exerts ty of food or the	animals, maint undings. luctive of the E its influence of e incidence of d ms are endog	cain a constant learth's biomes. In most insect posisease. In most insect posisease.	pody temperate pulations by a regulation by	false one (10 marks): ure irrespective to the ffecting the availability other environmental
 The organal similar Synchronal find and control find ov Cold hardal low ten The studyal Ecology 	b) differ ized eclosion in nate and food. iposition sites. liness refers to a aperature. by of the interrelacy ber of individua	I together in nate of the character of t	d. d) all abovemees to	ve mentioned	ng, must beeeeea particular time is the
 Define dia Describe to Define Ec 	e following Questipause and mentipe categories of otype giving another environments	tion its types. f migration in i example of the	nsects, explain	only one examptions that show	ple. ved how ecotypes
1-The ability	he following ser to extract nutr	ients from the	various textures	s is primarily d	lue toween

1969

TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF CHEMISTRY

EXAMINATION FOR SOPHOMORES (SECOND YEAR) STUDENTS OF BIOLOGY (SPECIAL BIOCHEM., CHEM./BIOCHEMISTRY, CHEM./ZOOLOGY AND CHEM./ENTOMOLOGY)

1969 COURSE TITLE:

INSTRUMENTAL ANALYSIS (1)

COURSE CODE: CH2244

DATE:

MAY 30, 2015

TERM: SECOND

TOTAL ASSESSMENT MARKS: 100

TIME ALLOWED: 2 HOURS

Question (I):

(25 mark)

A. Answer the following:

- 1- What is a photomultiplier tube? Describe what it does and how it works.
- 2- Most of phosphorescence cannot be recorded in solution at room temperature, why?

B. Choose the correct answer for the following:

1) Which of the fol	lowing is not active in IR abs	sorption spectroscopy?	
a) Cl ₂	b) CHCl ₃	c) CH ₄	d) C ₆ H ₆
O\ \A/L:-L:-			,

- 2) Which is the preferred continuum source in visible region?
 - a) Tungsten filament lamp b) Hollow cathode lamp c) deuterium lamp d) none of above
- 3) A photon whose wavelength is 200 nm is:
 - a) a visible photon b) an ultraviolet photon c) an infrared photon d) a microwave photon
- 4) The excellent exciting light source used in the atomic absorption spectroscopy is:
 - a) low pressure lamp b) hollow cathode lamp c) Xenon lamp d) tungsten-halogen lamp
- 5) Fluorescence occurs within:
 - a) 10 s

b) 10 ms

c) 10 us

d) 10 ns

Question (II):

(25 mark)

Answer the following:

- 1- List the following in order of increasing energy and wavelength: x-rays, infrared light, visible light, radio waves, and ultraviolet light.
- 2- What are an absorption filter and a monochromator?
- 3- Explain the energy transitions caused by UV-VIS light absorption.
- 4- Describe two light sources used for UV-VIS spectrophotometry.

Question (III):

(25 marks)

A. Discuss briefly on each of the following:

- 1- Hock's law and draw three examples of IR absorption modes.
- 2- Two applications of UV-VIS spectrophotometry?
- 3- Steps of atomization in the flame of atomic absorption spectroscopy.
- 4- Variation of fluorescence intensity with concentration.

Question (IV):

(25 marks)

Draw the following:

- b) Block diagram showing all the components of a basic spectrophotometer.
- c) Energy transitions involved in fluorescence and phosphorescence.

B. Answer the following problem:

What is the absorbance given that the molar absorptivity is 2.30×10⁴ L.mol⁻¹.cm⁻¹, the pathlength is 0.05 cm, and the concentration is 0.0000453 M?

Good Luck

Examiners	Prof. Dr. Mohamad Mohamad Ayad
	Dr. Nagy Labieb Kamal



Tanta University
Faculty of Science
Chemistry Department

Final Exam (Organic Chemistry 3) (CH2214)

2nd Level (Chemistry- Entomology) June 2105.

Time: 2 hours. Total marks: 150

Answer the following questions

1-Discuss the following:

[18 Points]

A) Diels-Alder reaction (give examples).

B) Claisen condensation reaction (explain the mechanism of the reaction).

C) Reformatsky reaction (explain the mechanism of the reaction).

2- Carry out the following conversions:

[12 Points]

- A) Benzaldehyde to cinnamic acid.
- B) Ethyl acetoacetate to 2-pentanone.
- 3- What statement about the aldol condensation is correct? [4 Points]

A) A Lewis acid is commonly used as a catalyst.

- B) The initial step is probably the formation of a carbanion.
- C) A Lewis base is employed to induce carbocation formation.
- D) The carbon chain is lengthened through the elimination of 1 mole of water.
- 4- Write an account on non-kinetic isotope effect.

[5 Points]

5- Discuss the mechanism of the following reactions:

[12 Points]

- A) Addition of HBr to 1,3-butadiene.
- B) Addition of HCN to methyl vinyl ketone.
- 6-Which of the following reactions exhibits primary kinetic isotope effect? (Explain your answer). [5 points]
 - A) E1 reactions.

B) E2 reactions.

- C) Nitration reaction of benzene.
- 7- Discuss the mechanism of the following reaction [5 points] 2,4-Dinitrochlorobenzene + NaOH \rightarrow 2,4-dinitrophenol.
- 8- Which of the following is the best leaving group in S_N 2 reactions?
 - A) C1
- B) Br
- C) I
- D) F

[5 points]

9 -Which of the following statements correctly describe(s) S _N 1reactions of alkyl halide (RX)? [5 points] I) Rate = k [base] II) Rate = k [base][RX] III) Rate = k [RX] IV) The reactions occur in two steps. V) The reactions occur in one step. VI) Rearrangements sometimes occur. A) II and VI B) I only C) I and III D) I and IV E) III, IV and VI.
10- The reaction of diethylmalonate with sodium ethoxide produces ethanol and a [4 points] i- free radical ii- carbocation iii- molecular species iv- carbanion v- carboxylate ion.
11 -Discuss S _N i mechanism (give example). [5 points]
12 -Arrange the following compounds according to their reactivity towards nucleophilic addition reactions to carbonyl group:[5 points] Acetone, Formaldehyde, Acetaldehyde.
13 -Discuss the mechanism of nucleophilic addition reactions to carbonyl compounds. [5points].
 14- Arrange the following compounds in the order of reactivity towards S_N2 reactions (explain your answer): [5 points] (A) β-Phenylisopropyl bromide (B) Benzyl bromide (C) α-Phenylethyl bromide.
 15-Which statement is correct for an S_N1 reaction at a chiral carbon atom? [5 points] A)The product will be optically active, but have opposite configuration. B)The reaction will involve racemation. C) A carbanion is formed as intermediate. D)The rate of reaction is a function of the concentration of the nucleophile. E)The attacking group will be a strong electrophile.
16-The addition of Br ₂ to trans-2-butene giving meso-2,3-dibromobutane can be explained by a mechanism involving: [5 points] A) A free radical. B) A carbocation. C) A cyclic bromonium ion. D) A carbanion. E) Simultaneous attack by bromine atoms.

17- Discuss the mechanism o	f Pinacol-Pinacolone rearrangement
reaction. [5 points]	
of alkyl halide (RX)? [:	II) Rate = k [base][KA] IV) The reactions occur in two steps. n one step.
be explained by a mecha A) a carbanion. C) a cyclic bromonium io	
20-Which two reaction type A) S _N 1 and E2 reactions C) S _N 2 and E1 reactions	es have the same initial step? [5 points] B) E1 and S_N1 reactions D) none of the above.
Chlorobenzene + KNF A) Aromatic electrophi B) Aliphatic Nucleophi	lic substitution.
22- Inversion of configurate A) E1 reaction C) S _N 1 reaction	tion is associated with which of the following? B) free-radical halogenation [5points] D) S _N 2 reaction E) none of the above.
hydroxide to 2-methyl	1-chloropropane with alcoholic potassium propene occurs through what mechanism?
[5points]. A) $S_N 1$ B) E1	C) E2 D) $S_N 2$ E) A and C.
24- Arrange the following	compounds in the order of reactivity towards [5 points] , CH ₃ CH ₂ CH ₂ CH ₂ Cl , CH ₃ CH ₂ CHClCH ₃ (B) (C)
25- Discuss the Free-ener [5points].	gy profile for a reaction with an intermediate.