

ISOLATION AND IDENTIFICATION OF THE
PHENOLICS FROM Ricinus communis L.

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ABSTRACT

The phenolic constituents of the leaves of Ricinus communis L. have been investigated by paper, thin layer and column chromatography. Spectral (IR and UV) analysis and high performance liquid chromatography (HPLC) were used in identifying the isolated components.

Nine compounds were characterized. Three of these compounds are recorded for the first time Viz isoquercetrin, 2,5-dihydroxy benzoic acid and (-)-epicatechin. The six known compounds are rutin, hepyroside, quercetin and chlorogenic, neo-chlorogenic and gallic acids. Hydrolysable and condensed tannins were also detected.

All the flavonoids isolated were of the flavonol-O-glycosides.

INTRODUCTION

Plants of the Euphorbiaceae are of considerable economic importance, and the products obtained from this family

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Isolation and Identification

include oils, various alkaloids and flavonoids (Rizk, 1968). Castor bean (Ricinus communis L.) is naturalized and is cultivated in Egypt for its medicinal importance.

The leaf of Ricinus communis L. contains rutin (Rizk, 1986). Moreover, the presence of quercetin and hyperoside in the leaf has been reported (Khafagy et al., 1986). Chlorogenic acid, neochlorogenic acid, gallic acid and an unidentified derivative of herbacetine were also isolated from the plant (Rizk, 1986).

Phytochemists and chemotaxonomists are attracted by the importance of secondary plant products of which phenolics are the most important. In a continuation of the phytochemical studies of Ricinus communis L. the isolation and identification of some phenolics from the plant leaves are reported.

MATERIALS AND METHODS

Plant material and germination

Castor bean seeds were germinated in ordinary garden soil,* in growth pots (Mayer and Shain, 1974). The pots were taken to a thermostatically controlled growth chamber

*CO₃=Zero, HCO₃= 0.7625 mwq/1, Total organic matter=9.92%, oxidizable organic matter= 2.250%, Bulk density= 1.074 gm/cm³, Porosity = 49% and Soil texture=% of sand=73%, silt=16%, clay=11%.

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adjusted to $25 \pm 2^\circ\text{C}$. After complete germination (15 days) the green leaves were obtained.

Extraction and isolation of phenolic compounds

The leaves were dried then powdered and defatted with ethyl acetate, extracted with methanol and concentrated. Hot distilled water was added to the extract and after cooling the aqueous/methanolic solution was filtered (Thieme and Khogali, 1975).

Flavonoids

Half the aqueous/methanolic extracts was concentrated and applied to silica gel column.

Phenolic acids, free aglycones and (-)-epicatechin

The other half was evaporated then subjected to continuous ether extraction. It was then shaken several times with water. The etherial layer contained the free aglycone.

Tannins

Tannins were detected in the aqueous extract of the leaves as described by Shmidt et al. (1955).

Standard methods of purification and identification of all compounds were followed (Mabry et al., 1970, Harborne, 1967, Markham, 1982) as well as by comparison with authentic samples. Also cochromatography with pure samples was employed. Acid hydrolysis was carried out using 2% H_2SO_4 for 120 min.

Paper chromatography (Pc)

Filter paper : Whatman No. 1 and 3

Developing solvent systems:

Flavonoids (Harborne, 1984)

BAW : n-Butanol-acetic acid-Water (4:1:5)

HOAc-H₂O : Acetic acid-water (15%)

PhOH-H₂O : Phenol-water (3:1)

Spray reagent : Zirconiumoxychloride (ZrOCl₂) 2% in methanol

Sugar moieties of flavonoids (Lewis and Smith, 1967)

BAW and

BPW : n-Butanol-pyridine-water (6:4:3)

Spray reagent : Anilin hydrogen phthalate.

Aglycone moieties

BAW and HOAc-H₂O.

Spray reagent: (ZrOCl₂)

Phenolic acids (Harborne, 1973)

BAW, HOAc-H₂O and

BEW : n-Butanol-ethanol-water (4:1:2.2).

Spray reagent: Ferric chloride (FeCl₃) 5% in methanol

Thin layer chromatography (TLC)

Adsorbent : Silica gel plates (20x20)

Developing solvent systems:

Aglycones

BPF : Benzene-pyridine-formic acid (36:9:6)

EtOAc-MeOH-H₂O : Ethyl acetate-Methanol-water (30:5:2)

Spray reagent : ZrOCl₂ and FeCl₃

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Phenolic acids

C-EtOAc-f : Chloroform-ethyl acetate-formic acid (5:4:1)
HOAc-C : Acetic acid-chloroform (1:9)
EtOAc-B : Ethyl acetate-benzene (9:11)
TCA : Toluene-chloroform-acetone (40:25:35)
Spray reagent : FeCl₃ (5%)

The ultra-violet (UV) and infra-red (IR) spectroscopic analyses were recorded by Perkin-Elmer lambda 48/Vis and Perkin-Elmer, recording IR spectrophotometer model 1430 respectively. For High performance liquid chromatography (HPLC) Perkin Elmer series 38 liquid chromatograph was used.

RESULTS AND DISCUSSION

Extraction of the dried leaves of Ricinus communis revealed eight phenolic compounds: four flavonoids, three phenolic acids and (-)-epicatechin. However, one of the flavonoids; isoquercetin and one of the phenolic acids; 2,5-dihydroxy benzoic acid, as well as (-)-epicatechin were detected for the first time.

Neochlorogenic acid which has previously been determined in Ricinus communis (Rizk, 1986) could not be detected under the prevailing experimental conditions.

The chromatographic data of the isolated compounds in different solvent systems are given in Table 1.

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Isolation and Identification

UV Spectra were measured after the addition of the following reagents separately to the methanolic solution of the flavonoids: aluminium chloride (AlCl_3), aluminium chloride/hydrochloric acid (AlCl_3/HCl), sodium acetate (NaOAc), sodium acetate/boric acid ($\text{NaOAc}/\text{H}_3\text{BO}_3$). The absorption maxima are shown in Table 2.

Isoquercetin (quercetin, 3-glucoside) was identified by its chromatographic and spectral analysis. Chromatographic investigation of the separated aglycone showed that it possesses the same R_f value and UV-spectral analysis as quercetin. It was also undepressed upon admixture with authentic quercetin. The R_f values of its sugar moiety were identical with glucose (Table 1).

The infra-red spectrum of 2,5-dihydroxybenzoic acid revealed the presence of a phenyl group in the range $723-756 \text{ cm}^{-1}$, hydroxyl groups at 3121 cm^{-1} and carbonyl group at 1752 cm^{-1} . The IR-spectral analysis of the identified phenolic acids are shown in Figs. 1-3.

(-)-Epicatechin is detected for the first time in R. communis. It was characterized by its chromatographic data and IR-spectral analysis (Tab. 1 and Fig. 4).

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Flavonoids, phenolic acids and (-)-epicatechin were also identified by HPLC, using pure standard for qualitative comparison (Figs. 5 and 6).

Both hydrolysable and condensed tannins were detected in the leaves of R. communis. The presence of the two groups of tannins has not been reported previously.

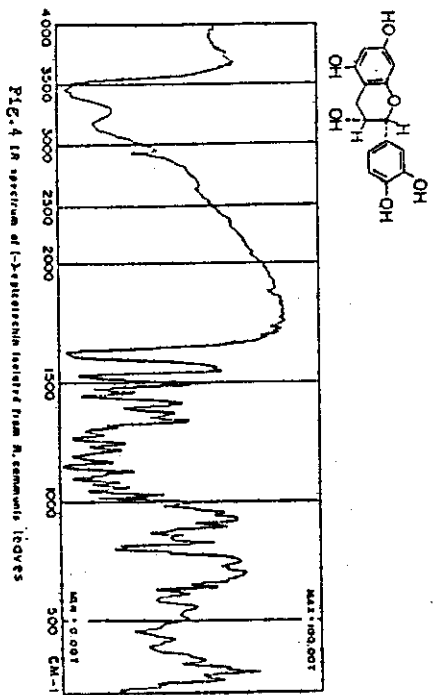
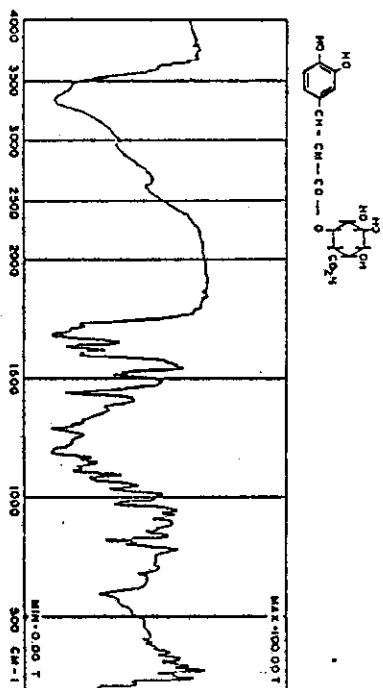
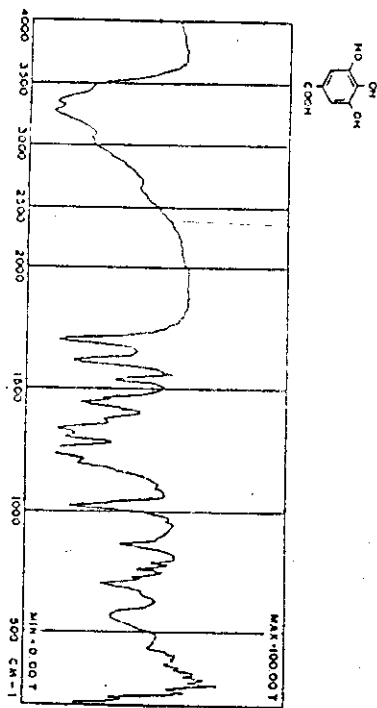
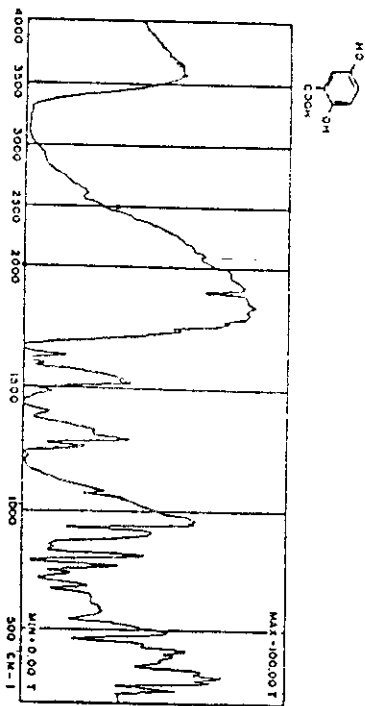
The data obtained, so far, show that all the glycosidic patterns are flavonol-O-glycosides of quercetin. The different phenolic patterns are related to their occurrence to the family Euphorbiaceae.

Table 1. R_f values of the phenolic compounds isolated from the leaves of *R. communis*.

Compounds	PC			R _f (x100) in			ILC				
	9AW	H ₂ O	PhOH	BEU	9PW	BPf	EtOAc-MeOH-H ₂ O	C-EtOAc-F	HOAc-C	EtOAc-B	TFA
<u>flavonoids</u>											
Quercetin (Q)	76	84			16	87					
Q-3-glucoside	71	39	54			63					
Q-3-galactoside	70	37	56			65					
Q-3-rutinoside	57	60	45			35					
<u>Phenolic acids</u>											
2,5-dihydroxybenzoic acid	77	73		50				33		45	
3,4,5-trihydroxybenzoic acid	62	50						32		39	
chlorogenic acid	73	65		45				09			15
<u>(-)-Folicatechin</u>	65	55	59								
<u>Sugar alcohols</u>											
Glucose	18			50							
Galactose	16			44							
Thamnose	30			70							

Table 2. UV-spectrum of the flavonoids in *Ricinus communis* leaves!

Flavonoids	MeOH	Spectral maxima										
		NaOMe	AlCl ₃	AlCl ₃ /HCl	NaOAc	NaOAc/H ₂ BO ₃						
Quercetin (Q)	373	323	220	455	223,270	423	225,270	405	227,276	402	227,262	
Q-3-glucoside	368	430	227,279	353,405	229	343,407	227	227	403	229,252	388	229,261
Q-3-galactoside	360	257	410	270	440	275	430	270	380	270	380	260
Q-3-rutinoside	348,370	249,259	322,413	226,272	366,402	225,266	355,404	225,267	403	226,273	393	227,263



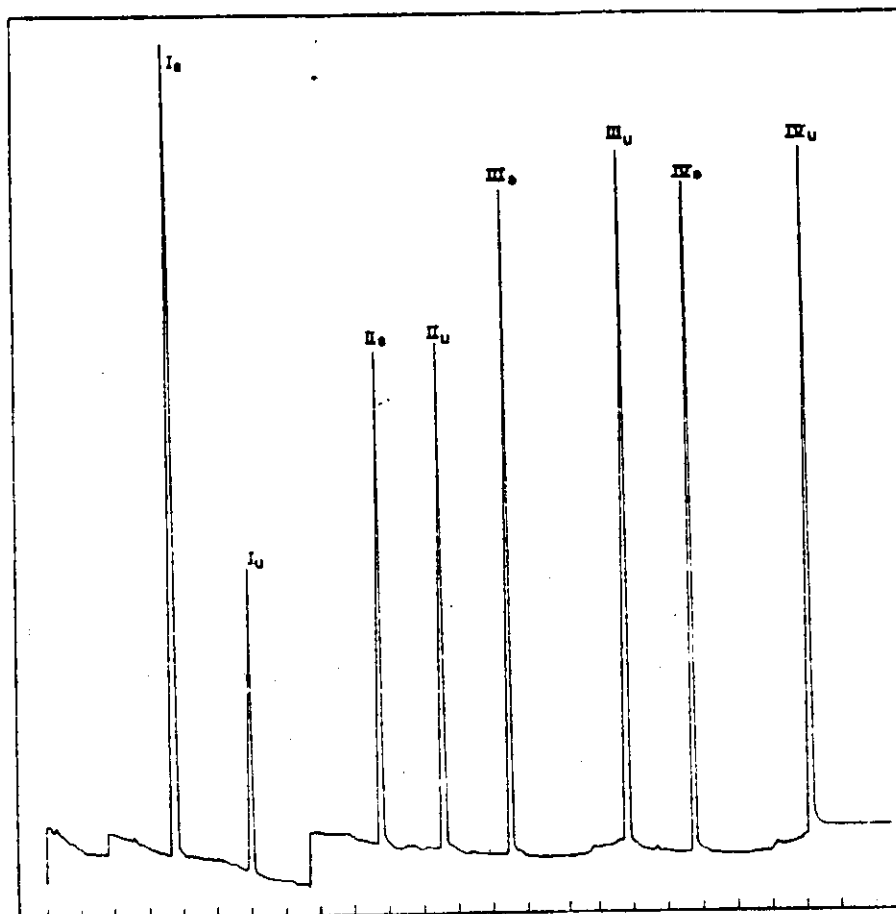


Fig. 5: HPLC - analysis of Flavonols isolated from *R. communis* leaves

I - Rutin ; II - Quercetin ; III - Isoquercitrin ; IV - Hyperoside

S - Standard ; U - Unknown

Retention time $t_R = 6$ min. , Perkin Elmer series 3B liquid chromatograph.

Column analytical HC - ODS , Solvent 1% acetic acid : methanol (9:1).

rate of flow 2 ml/min. ; temp. = 24°C , detection : UV detector .

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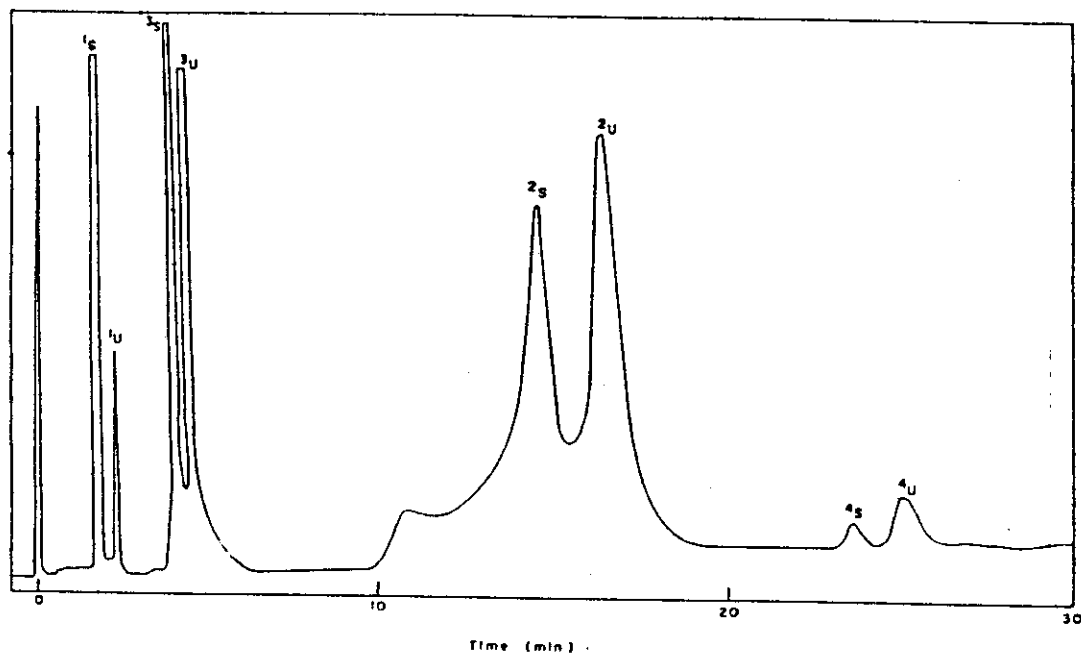


Fig.6 HPLC - analysis of phenolic acids and (-)-epicatechin isolated from *R. communis* leaves

S = Standard ; U = Unknown

t_R = Retention time (min) , Column analytical MC - ODS , Solvent 1% acetic acid : methanol (9:1) ,
rate of flow 2 ml/min ; temp = 24 °C , detection : UV detector

Compound	t_R	
	S	U
1- Gallic acid	2.16	2.18
2- 2,5 - dihydroxy benzoic acid	15.89	16.16
3- Chlorogenic acid	4.26	4.28
4- (-) Epicatechine	24.51	25.60

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استخلاص وتعريف المواد الفينولية
من نبات الخروع R. Communis

× عائشه خوجلى - ×× صلاح الدين يس بركات - ×× حنان ابو زيد
× قسم التاريخ الطبيعى - كلية التربية - جامعة الاسكندرية
×× قسم النبات والميكروبيولوجى - كلية العلوم - جامعة الاسكندرية

يتضمن البحث فصل وتعريف المواد الفينولية الموجودة فى اوراق نبات
الخروع Ricinus communis بعد انبات البذور فى درجة حرارة ٢٥ م. وقد
استخدم فى فصل وتعريف هذه المركبات طرق كروماتوجرافيا الورق والطبقة
الرقيقة واعمدت الفصل بجانب وسائل الامتصاص الطيفى للأشعة تحت الحمراء
وفوق البنفسجية او طريقة الكروماتوجرافى تحت الضغط العالى HPLC . وقد
اوضحت هذه الدراسة ان اوراق نبات الخروع تحتوى على عدد من المواد الفينولية
والتي قسمت حسب تركيبها الكيميائى الى : مواد فلافونيدية :
(Hyperoside, isoquercetin* quercetin, rutin) واحماض فينولية
(الجاليك ، الكلوروجينيك ، *آره ثنائى هيدروكسى بنزويك) والمركب ابيكاتشين
(*epicatechin) الذى يعتبر المادة الاولية لتكوين التانينات . ثلاثة من
هذه المركبات تعرف لأول مرة فى بادرات الخروع وهى التى تحمل العلامة * .
وكذلك امكن الاستلال على وجود مجموعة التانينات المكثفة والمجموعة القابلة
للتحليل .

وقد بينت الدراسة ان الكورستين يوجد فى صورة حرة والجليكوسيدات
المتصلة به تنتمى الى مجموعة فلاونول - 0 - جليكوسيد .