



المستوى الثالث كيمياء/ ميكرو eyel l'adi



TANTA UNIVERSITY FACULTY OF SCIENCE BOTANY DEPARTMENT



امتحان الفصل الدراسي الثاني للمستوى الثالث كيمياء/نبات

Course Title:	E	Course Code: Bo 3206	
June 6, 2017	Term: Second	Total assessment marks: 100	Time Allowed: 2hour

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				ل: ضع علامة (\lor) أو علاما	
) (آ درجات)	ن الحولية (ها غنية بالنباتان	مائص النباتات المصرية كونه	۱ ـ من خص
	(٦ درجات)	لينيس ()	ن ۱۷ علی ید	سمية الثنائية للنباتات في القر	٢- بدأت الد
	•	(۲ درجات)	() 5	قة البحر الأحمر غنية بالفلور	۳- تعد منط
	a n	(۱ درجات)	()	بل علبة بالصحراء الشرقية	٤ - يوجد جا
) (آ درجات)	جغرافية معينة (بة نشاتات منطقة	سائص التقسيمي	العام يشتمل على سجل بالخص	٥- المؤلف
		i i	۳ درجة)	ي: أكمل العبارات التالية (.	السوال الثان
(٦ درجات)				عيرة البردويلُ في	١- ئوجد بد
(٦ درجات)		و		نباتات الملحية إلى	٢- تنقسم ال
(٦ درجات)		و	وية	ات المميزة للمناظق الصحرا	٣- من البيد
(٦ درجات)		وو		عان من المكابس هي	٤- يوجد نو
(٦ درجات)		و		ع المفاتيح التعريفية	٥- من أنوا
	9	20	(ä	ث: ناقش کلا من: (۳۰ درج	السؤال الثال
	رجات)	ت فقط) (۱۰ در	(أذكر ٥ بيئا	لمختلفة في منطقة دلتا النيل	ا - البينات ا
		a j	(۱۰ درجات)	يوة وواحة وداي النطرون	٢- واحة س
	8 8		(۱۰ درجات)	بال علبة	٣- منطقة ج

السؤال الرابع: أجب واحدا من الأسئلة التالية (١٠ درجات) ١- قمت برحلة إلى مرسى مطروح، وقمت بتجميع عينات بياتية ، كانت احداها عينة عصيرية وضح كيفية عمل عينة معشبية منها

٢- خصائص الفلورة المصرية

تمنياتي بالتوفيق والنجاح

Examiners: Dr. Ahmed Sharaf El-Din and Dr. Dalia Abd El-Azeem Ahmed لجنة المصححين أ.د. أحمد شرف الدين عبدالوهاب - أ.د.داليا عبد العظيم أحمد – أ.د. عاطف أبو شادي

ALC: 1		7.	Tanta University	-			
M			Faculty of Science				
			Department of Chemistry				
16	Final exam. for Juniors students of doubled branches						
1969	Course title: Molecular Photochemistry Course Code:						
Date:	May 30, 2017	Term: second	Total assessment Marks: 50	Time allowed: 2 H			

Answer the following questions

1) Differentiate between each of the following:

(16 marks)

- a. El-Sayed's rule and Kasha's rule
- b. E- and P- types of delayed fluorescence
- c. 1,2 addition and 1,2-1,4 addition of photodimerization of olefinic compounds.
- d. Radiative natural and observed fluorescence lifetimes.
- 2) Draw each of the following:

(12 marks)

- a. Triplet-triplet mechanism of energy transfer action according to the Dexter mechanism.
- b. Possible transitions causing energy transfer processes in Biacety, Pyrene and Naphtalene system.
- c. Singlet-singlet mechanism of energy transfer action according to Förster mechanism.
- d. Jablonski Diagram for electronic transitions between ground and excited states.
- 3) Write down the following statements. Show whether each of the following statements is true or false, if false, please, write down the true. (14 marks)
- a. The energy gap value, $\Delta E(T_1 \sim S_0)$ is a factor which determines the relative magnitudes of k_f and k_{isc} ($S_1 \sim T_1$).
- b. Each decay process represented by $k_{\rm f}$, $k_{\rm ic}$, $k_{\rm r}$ and $k_{\rm et}$ is bimolecular rate constant.
- c. Promotion of an electron to an antibonding molecular orbital upon excitation takes about (10⁻¹⁰ –10⁻¹²s), which is very quick compared to the characteristic time for molecular vibrations (10⁻¹⁵ s).
- d. The rate of fluorescence can be enhanced relative to the other processes by using heavy atoms.
- e. Excimers are dimers in the excited state. They are formed by collision between two excited molecules.
- f. Intersystem crossing (ISC) is an iso-energetic radiationless transition between two electronic states of same multiplicity.
- g. In Jablonski diagram, the triplet state(↑↓) is always of lower energy than the energy of the corresponding singlet state(↑↑).
- 4) Give short notes on the following:

(8 marks)

- a. Quantum yield of fluorescence and of phosphorescence, $\Phi_{
 m f},\Phi_{
 m p}$.
- b. Wigner spin conservation rule.

Good Luck

The examiners: 1. Prof. Dr. Samy el-Dally

2. Prof. Dr. Shakir T. Abdel-Halim

ALE I			Tanta University	46			
			Faculty of Science				
		DOTAL CONTRACTORS	Department of Chemistry				
	Final exam. for Juniors students of doubled branches						
1969	Course title:	Mole	cular Photochemistry	Course Code: CH3244			
Date:	May 30, 2017	Term: second	Total assessment Marks: 50	Time allowed: 2 H			

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(16 marks)

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2. Prof. Dr. Shakir T. Abdel-Halim

AE I			Tanta University				
20			Faculty of Science				
			Department of Chemistry				
1/6	Final exam. for Juniors students of doubled branches						
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Date:	May 30, 2017	Term: second	Total assessment Marks: 50	Time allowed: 2 H			

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 m f},\Phi_{
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Good Luck

The examiners: 1. Prof. Dr. Samy el-Dally

2. Prof. Dr. Shakir T, Abdel-Halim



DATE: 6/6/2017

COURSE TITLE

a) Agriculture

b) Sewage

JUNE, 2017

TANTA UNIVERSITY, FACULTY OF SCIENCE, DEPARTMENT OF BOTANY FINAL EXAMINATION FOR THE THRID YEAR (CHEMISTRY / MICROBIOLOGY)

SOIL AND WATER MICROBIOLOGY

TOTAL ASSESSMENT MARKS: 100

COURSE CODE: MB3202

TIME ALLOWED:2 HOURS



Answer the following question

	D	mower the to	nowing question		
1.	Compare between the following	19		2 M	(20 marks)
	 Endomycorrhiza and Ectomycor Protocooperation and mutualism BOD and COD Fecal and non-fecal coliform 	rhiza	(a. W.) p	1,,51	
2.	. Mention the following		er in a serie of the series		(25 marks)
	 The negative associations betw Advantages of the MF Techniqu Regular pattern of distribution o The benefit of both fungi and high Disinfectants and potential Hear 	een soil microor ie. f soil microorgar gher plant from	ganisms nisms mutualistic association		The second second
3.	. Check $\sqrt{\text{ or X for the followin}}$	g sentences			(10 marks)
	1. A drinking-water distribution syst	em provides a h	abitat for microorganisi		
	2. The surface layer of soil contains	lower number	of microorganisms		- ()
	3. Soil air serves as a solvent and o	carrier of nutrien	ts		(,)
	4. Autotrophs are "primary consum	er			()
	5. Protozoa add organic matter to s	oil when die and	d thus increase the amo	ount of organic carb	on ()
	6. The water is protected from reco	ntamination dur	ing distribution by free	chlorine	()
	7. The presence of combined chlor	ine in water is co	orrelated with the abser	nce of microbes	
	8. Industrial sewage represents an	excellent mediu	m in which to grow bac	teria and microbes	()
	9. Water-based diseases include d	engue and mala	ria		()
	10. Water-washed diseases is direct	ctly related to wa	ater supply or quality		()
4.	. Chose the correct answer		e were de regalitée de perféquipe		(20 marks)
	Which of the following is:	n an example of	Water-related disease	s de la companya de l	Haraki e
			d) dysentery d) All of t		
	2. b) Decrease in light and	,			

c) Shipping

d) All of these

).	The Positive associations	between soil micro	organisms are			
			a) The amount of oxygen b) The amount of oxygen c) All of the above		tem by respiring a	• •		
	2	١.	The absence of disease-	causing organisms	is correlated with	the presence of		
			a) Residual chlorine		b) Free c	hlorine		
			C) Combined chlorine		d) none d	of the above		
	;		The most common pathod to	gens in surface wate	er are Cryptospori	dium and Gardi	a Lamblia, belongs	3
			a) Fungi c)	Viruses	c) Bacteria	d) None o	of the above	
	63	3.	Microbial examination of	water by using the	indicators microor	ganisms is	W.	
			a) Time consuming		b) Expensive		
			c) Potentially dangerous	to lab personnel	- d) None of the at	ove	
		7.	Different soil microorgani	sms, having soil bin	ding properties ar	e graded in the	order as	
			a) fungi > Actinomycetes	s > yeasts	b) fungi > yeasts	s > Actinomycete	es	
	90 59		c) Actinomycetes > fung	i > yeasts	d) yeast > Actino	omycetes > fung	ji	Ti.
		8.	Purification of water cont	aining the following	steps in the order	as		
			a) Aeration, Flocculation,	Sedimentation, Filt	ration and Disinfe	ction		
			b) Aeration, Filtration, Se	dimentation, Floccu	lation and Disinfe	ction		
			c) Aeration, Sedimentation	on, Flocculation, Filti	ration and Disinfe	ction	_ a 8 av	
		9.	Nutrients and Pesticides	are pollutants resu	Iting from	× ⁸		
2		1	a) Agriculture b)	Aquaculture	c) Sewage disch	narge d) None	of the above	
		10.	Microorganisms may be	introduced into the	distribution systen	n during		
			a) Drilling b) servicing	c) repairing	d) All of	these	
5. G	ive re	eas	on(s) for			8 8	(25 mark	s)
a To so He D	š.,	1- 2- 3- 4- 5-	MF technique is not suit The most commonly use	ole as disinfectant the able for water samp ed indicator microor	lles which contain ganisms are colife	many impurities orm bacteria	S	

With my best wishes



TANTA UNIVERSITY FACULTY OF SCIENCE DEPARTMENT OF BOTANY

EXAMINATION FOR THIRD YEAR STUDENTS OF MICROBIOLOGY SPECIAL

ENERGY BIOCONVERSIONS & PHOTOSYNTHESIS COURSE TITLE: COURSE CODE: 3220 TERM: SECOND TOTAL ASSESSMENT MARKS: 150 DATE: 1-6-2017 TIME ALLOWED: 2 HOURS

Q1. Define (8 marks; 2 each): uphill reaction, phyeophytin, reaction center, quantum yield of photosynthesis

00		2.4		541 6	141	-0	920 1201		gs 33			
Q2.	Choose	the	letter	that	shows	the	right a	nswer	(10)	Marks:	1	each):

O2. Choose t	he letter that sho	ws the right ans	wer (10 Mar	eke 1 aaah)
	ated to two protein		wei (10 mai	KS, I cacil).
	(b) PsaA& PsaC		D ₂ (d) Psa	aA& PsaB
	of the biological s			
	(b) entropy		gy (d) usa	ble energy
	fer in the antenna	***		
(a) small	(b) reversible			versible
4. PSII& PSI ar	e located at			
(a) same protein	n (b) separatly (c)) chloroplast enve	elope (d) A	TP synthase
5. Plastocyanin	transfers electrons	s from PSII to		e 10
(a) P ₇₀₀	(b) pheophytin	(c) Q _A	(d) Q_B	
6. Chla fluoresc	ence is seen at (a) blue (b) red	(c) yellow	(d) green
7. If $\triangle G$ is positive.	ive, free energy is	±		
(a) released	(b) required	(c) consumed	(d) const	ant
8. Excited Chl*	can return to its g	round state in a p	rocess called	d
(a) fluorescence	(b) energy trans	sfer (c) heat	(d) al.	l correct
9. In non-photo	chemical quenchir	ng, excess light en	nergy is dissi	ipated as
(a) uphill reaction	on (b) Calvin cy	cle (c) heat	(d) PSI hyd	rolysis
10. Phyeophytir	accepts electrons	from		
(a) vitamin K ₁	(b) P_{700}	(c) ATP syntha	se (d)	P ₆₈₀

Q3. Compare between (24 Marks; 8 each):

- 1. Structure of Cyt b_6 f complex and the electron carrier plastocyanin
- 2. Components of reaction centers and light antenna complexes
- 3. Importance of Hill reaction and red drop & Emerson enhancement

Q4. Complete the following statements (34 Marks; 2 each):

- 1. In Q-cycle, transfer of 2 e to Q_B reduces it to ..., it takes 2H⁺ from ...yielding
- 2. The first photochemical reaction occurs when e is transferred from to
- 3. Exposure of chloroplast to excess energy causes
- 4. Energy transfer is a process while electron transfer is a process
- 5. protein is the main target of damage of photoinhibition
- 6. Photolysis of 2 water molecules produces, and
- 7. In thylakoid membranes, diffusion of protons from to causes the synthesis of one ATP by a complex called, in a process called

Q5. Answer the following question:

- 1. How Cam plants avoid photorespiration?
- 2. What are the possible fates of triose phosphates following their synthesis by the Calvin cycle?
- 3. In the processing of 2-phosphoglycolate, what are the significant steps that occur in the mitochondrion and the peroxisome? Where is ATP used? What is 2-phosphoglycolate processed into?

Q6. Complete the following sentences:

- 1. The carboxylation in C4 plants is catalyzed by......and occurs in......
- 2. Light compensation point is
- 3. The term green house effect refers to......
- 4. In Calvin cycle, 3-phosphoglecerate is reduced to.....through the use of.....generated by.....

Best Wishes

Prof. Dr. Fatma Elshintinawy Prof. Dr. Soad Elfiky

Markey Mass